Name: \_\_PARTIAL KEY\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Nomenclature

Binary Compounds and Compounds with Polyatomic Ions

1. (NH4)2SO4 11. K2Cr2O7

ammonium sulfate potassium dichromate

1. CuSO3 12. ZnBr2

1. Ag2O 13. Sn(NO3)4

silver oxide tin (IV) nitrate

1. Ag3PO4 14. KOH

1. SnO 15. Hg2SO4

tin (II) oxide mercury (I) sulfate

1. NH4OH 16. PbCrO4

1. CO2 17. FeCl3

carbon dioxide iron (III) chloride

1. ZnS 18. NaOH

1. NaClO2 19. KMnO4

sodium chlorite potassium permanganate

1. HgI2 20. PCl3

Name: \_\_PARTIAL KEY\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Formula Writing

Binary Compounds and Compounds with Polyatomin Ions

1. Potassium iodide 11. Lead (II) acetate

KI Pb(C2H3O2)2

1. Silver chloride 12. Barium oxide

1. Aluminum hydroxide 13. Calcium oxalate

Al(OH)3 CaC2O4

1. Chromium (III) chloride 14. Tin (IV) phosphate

1. Copper (II) oxide 15. Zinc carbonate

CuO ZnCO3

1. Mercury (II) bromide 16. Mercury (I) sulfate

1. Sodium chlorate 17. Carbon monoxide

NaClO3 CO

1. Dihydrogen monosulfide 18. Magnesium nitrate

1. Ammonium sulfite 19. Copper (I) sulfate

(NH4)2SO3 Cu2SO4

1. Silver chromate 20. Iron (II) sulfide